

## Tellurium(-II)

Reaction	Filella and May, 2019 <sup>a</sup>
$\text{Te}^{2-} + \text{H}^+ \rightleftharpoons$	11.81
$\text{HTe}^- + \text{H}^+ \rightleftharpoons \text{H}_2\text{Te}$	2.476

<sup>a</sup>The number of significant figures are retained to minimise propagation of round-off errors; they should not be taken to indicate the relative uncertainty of the values, which is always at least one order of magnitude less than indicated.

M. Filella and P.M. May, The aqueous chemistry of tellurium: critically-selected equilibrium constants for the low-molecular-weight inorganic species. *Environ. Chem.* 16, 289–295 (2019). doi:10.1071/EN19017